

Varme

Varmekapacitet

$$C = M \times c$$

$$E = C \times \Delta T$$

$E = \text{Energi}$

$M = \text{Masse}$

$\Delta T = \text{\AA}ndring i temperatur$

$C = \text{Celsius}$

$$[C] = J/K$$

Smeltevarme

$L_s = \text{den specifikke smeltesvarme} - 334 \frac{KJ}{Kg}$

$$E = L_s \times \Delta m$$

Fordampningsvarme

$$E = L_f \times \Delta m$$

$L_f = \text{den specifikke fordampningsvarme} - 2257 \frac{KJ}{Kg}$

$$[L_f] = \frac{kJ}{Kg}$$